

Methane in water and bottom sediments of Onego Lake

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The Greenhouse Effect

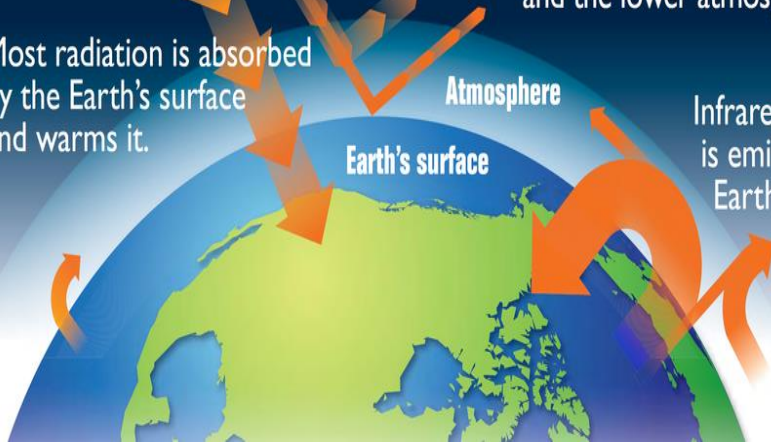
Some solar radiation is reflected by the Earth and the atmosphere.

Some of the infrared radiation passes through the atmosphere. Some is absorbed and re-emitted in all directions by greenhouse gas molecules. The effect of this is to warm the Earth's surface and the lower atmosphere.

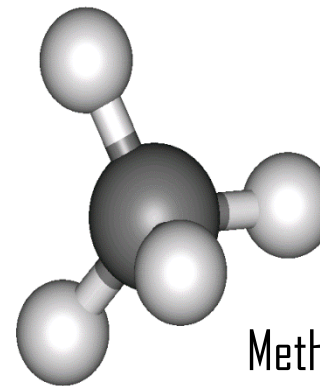
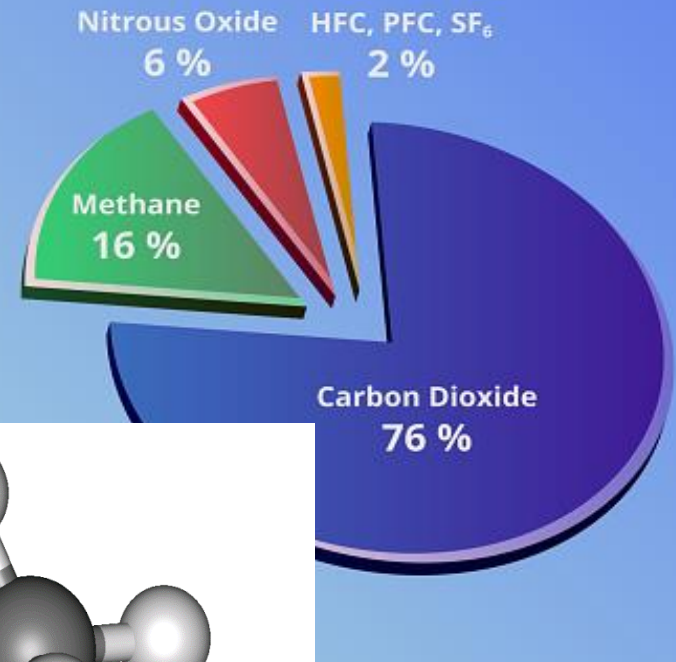
Most radiation is absorbed by the Earth's surface and warms it.

Atmosphere
Earth's surface

Infrared radiation is emitted by Earth's surface

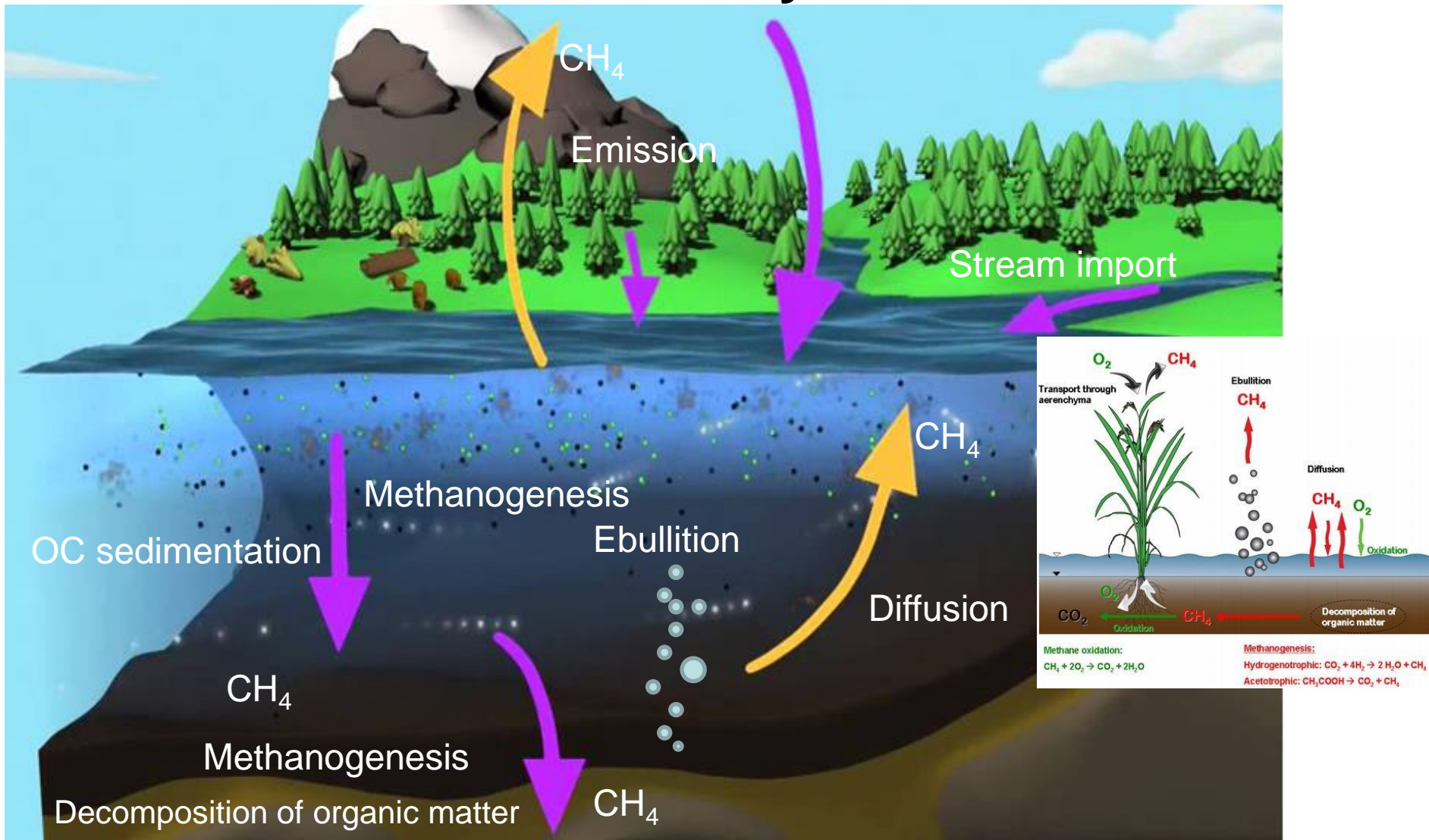


Greenhouse Gases



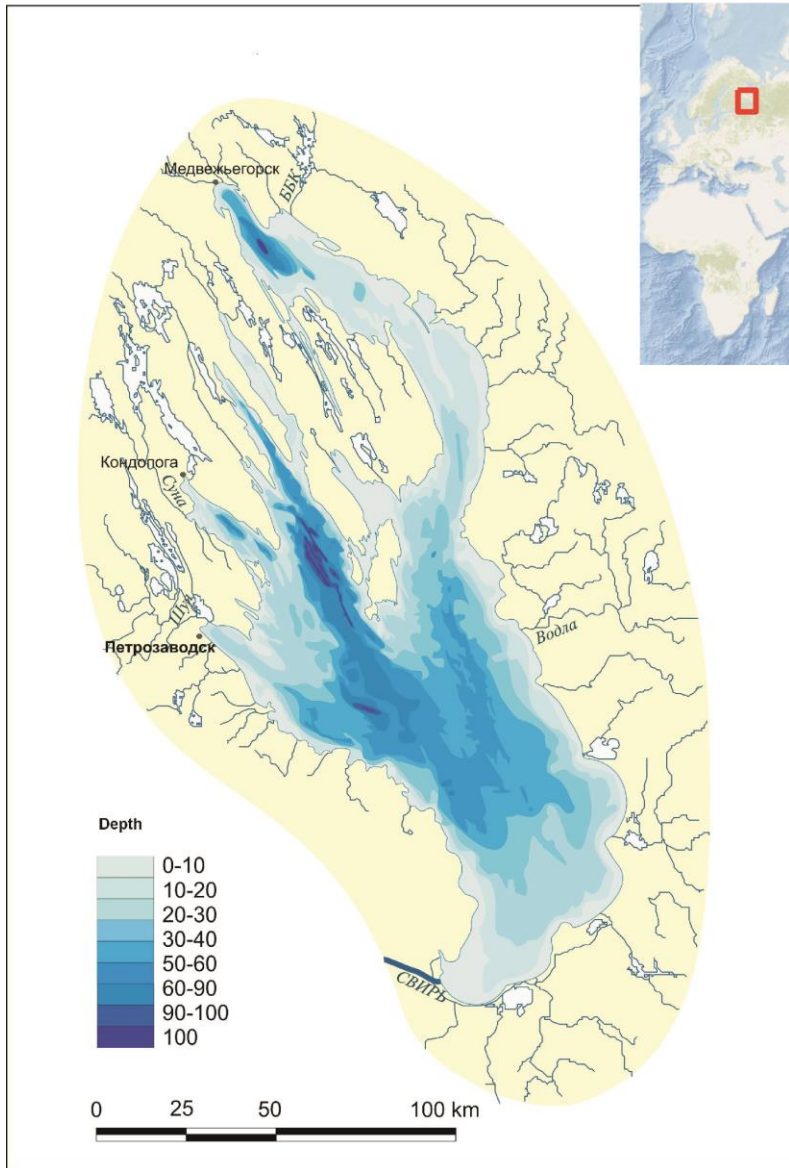
Methane is a greenhouse gas, accounting for about 20 percent of global emissions. Methane is more than 25 times as potent as carbon dioxide at trapping heat in the atmosphere. Over the last two centuries, methane concentrations in the atmosphere have more than doubled, largely due to human-related activities. Because methane is both a powerful greenhouse gas and short-lived compared to carbon dioxide, achieving significant reductions would have a rapid and significant effect on atmospheric warming potential.

Methane cycle

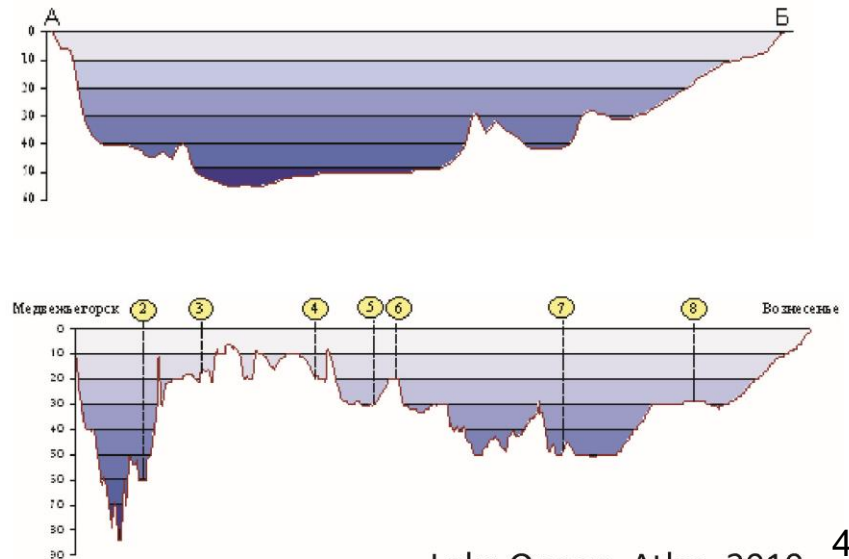


- The methane cycle is a geochemically important link in the global cycle carbon. The main processes of carbon transformation in this cycle are exclusively specific groups of microorganisms. Major pathways of CH₄ emissions include: methanogenesis via efflux through plant aerenchyma, ebullition, and diffusion.

Onego Lake is the second-largest lake in Europe



Surface area, km ²		9720
Shoreline, km		1810
Volume, km ³		295
Length, km		290
Width, km	mean	33.5
	maximum	82
Depth, m	mean	30
	maximum	120



Methane in Petrozavodsk bay

The results of long-term observations of the methane content in water and bottom sediments of the coastal areas of the Petrozavodsk Bay of Onego Lake are presented. The influence of various natural and anthropogenic factors on the formation and distribution of methane concentrations is considered.

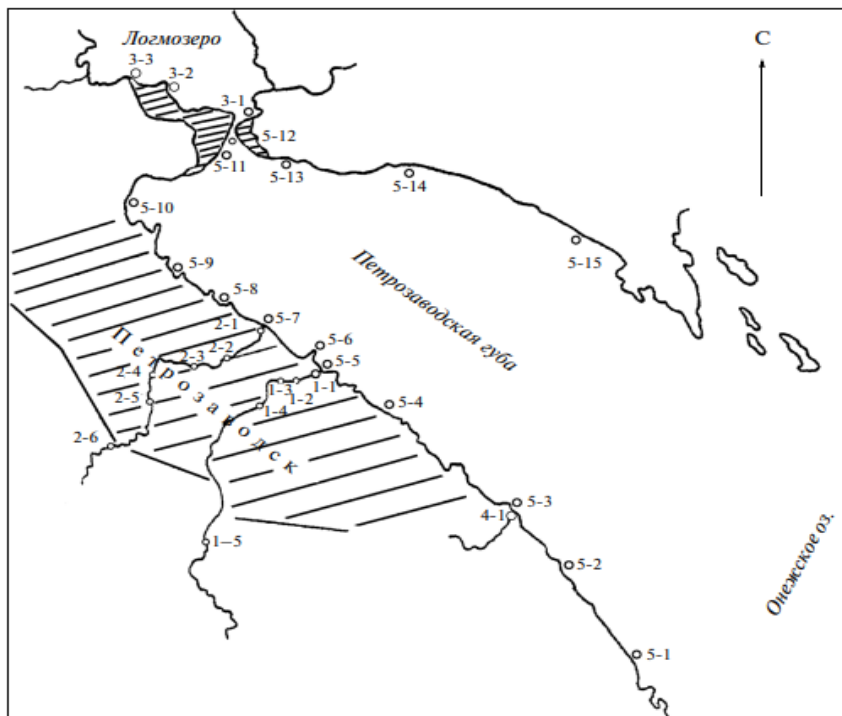


Рис. 1. Карта-схема расположения станций отбора проб воды и ДО в Петрозаводской губе Онежского оз. и в его при-токах в июне–июле 2007–2010 гг.

- Garkusha D.N., Fedorov Yu.A. Features of the distribution of methane content in the initiative areas of the Petrozavodsk Bay of Onego Lake // Water Resources, 2015. V. 42. No. 3. P. 331-339

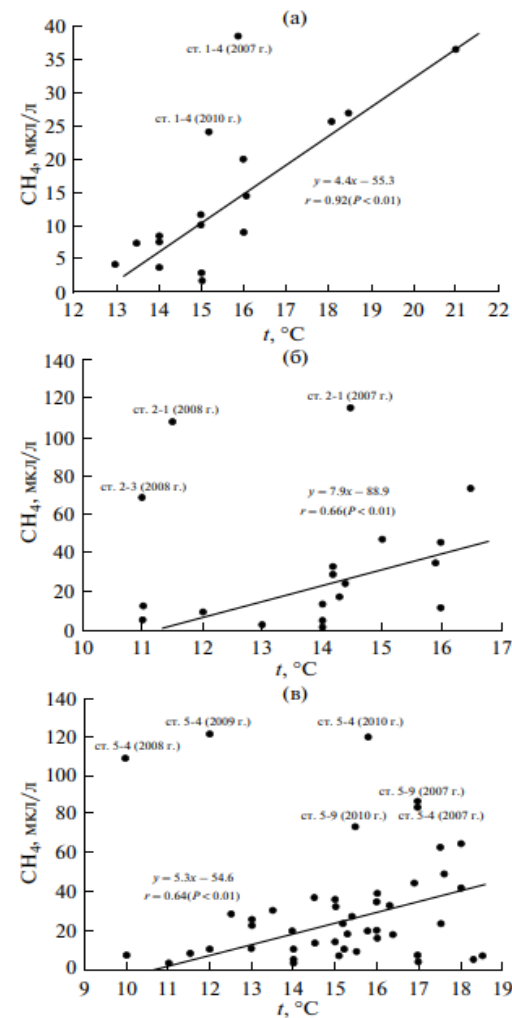
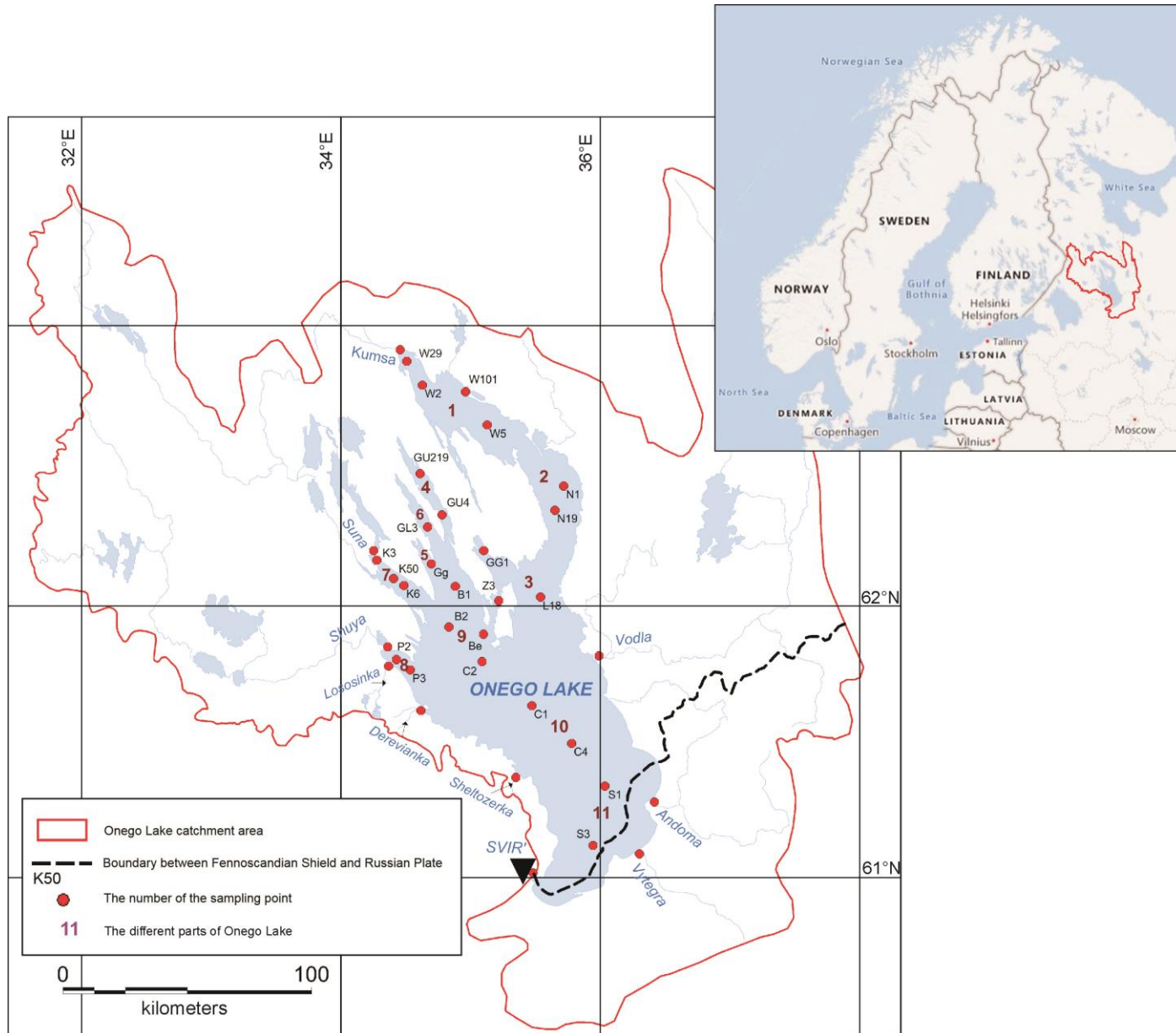


Рис. 3. Зависимости между содержанием метана и температурой воды, по данным 2007–2010 гг. а – р. Лососинка, б – р. Неглинка, в – Петрозаводская губа.

The Lake Onego sampling scheme



Sampling equipment

Bathometer



Gravity corer «LIMNOS»



Gravity corer with a plastic tube 50 mm in diameter, 1 m long



Gravity corer 127 mm in diameter, 3 m long



Echosounder



Peterson grab
volume 0.025 m²



Research methods

Methane analysis was carried out by the gas chromatographic method using the analysis of the equilibrium vapor of the mass fraction of methane in water and bottom sediments on a gas-liquid chromatograph Crystal 5000.1, Khromatek

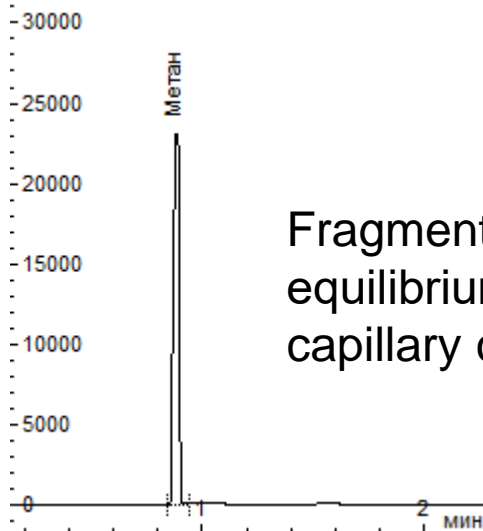


Regression equation
[Garkusha et al.,
2016]

- $\lg F_{\text{CH}_4} = 1,03891 \lg C_{\text{CH}_4} \text{ д. о.} + 1,4631$ ($r = 0,84$; $n = 46$; $P < 0,01$),
- where $\lg F_{\text{CH}_4}$ is the logarithm of the methane flux from sediments to water, $(\text{nl}/\text{m}^2)/\text{day}$; $\lg C_{\text{CH}_4}$ – logarithm of methane concentration in sediments, nl/dm^3 of wet sediment

Fick's law

- $J = -D \cdot dc/dx$,
- where J is the amount of a substance passing perpendicularly through a unit of surface area per unit of time, m^2/s of a wet sample; D is the diffusion coefficient, m^2/s (the self-diffusion coefficient was used - $0.188 \text{ cm}^2/\text{s}$, Sherwood et al., 1975); dc/dx – concentration gradient



Fragment of the calibration chromatogram of the equilibrium vapor of methane 200 μl on a capillary column DB-WAX 30m, 0.32mm, 0.50 μ



Research methods

Bottom sediments

- Eh - Potentiometric method with Pt electrodes and mediator Trilon-B.
- Loss on ignition (LOI) - Gravimetric method $T=550\text{ }^{\circ}\text{C}$
- Natural humidity ($W_{20\text{ }^{\circ}\text{C}}$) - Gravimetric method $T=20\text{ }^{\circ}\text{C}$
- Porosity (Por), specific gravity (m_{sp}), absolute humidity ($W_{150\text{ }^{\circ}\text{C}}$) - Gravimetric method $T=105\text{ }^{\circ}\text{C}$
- Ash content - Gravimetric method $T=900\text{ }^{\circ}\text{C}$
- C_{org} - Direct combustion of solids, TOC-V Series Total Carbon Analyzer with SSM-5000A Attachment
- Consumption of O_2 by sludge - Exposure during the day, Winkler bottle method
- Plant pigments - Extraction with acetone, spectrophotometric method $\lambda=663, 645, 630, 750\text{ nm}$
- Humic and fulvic acids - Photometric method after alkaline extraction, $\lambda=330, 360\text{ nm}$
- $N-NH_4^+$ - Conway microdiffusion method, titrimetric termination (NaOH)
- N_{org} - Kjeldahl method, Conway method
- Phosphorus mineral (P_{lab}) - Boiling with $5N\text{ H}_2\text{SO}_4$, photometric finish
- P_{total} - Kjeldahl Method
- Granulometric analysis - Mie and Fraunhofer laser diffraction combined with Polarized Light Differential Intensity Recording (PIDS)



TOC-V Series Total Carbon Analyzer with SSM-5000A Attachment, Shimadzu

Research methods

Water



GFA-X7, Shimadzu

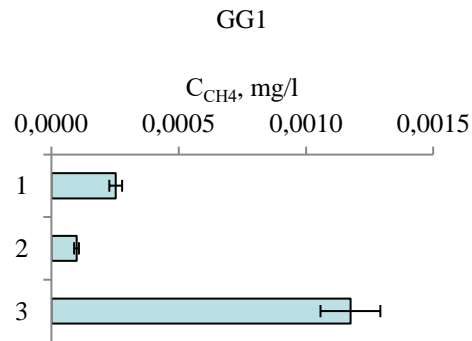
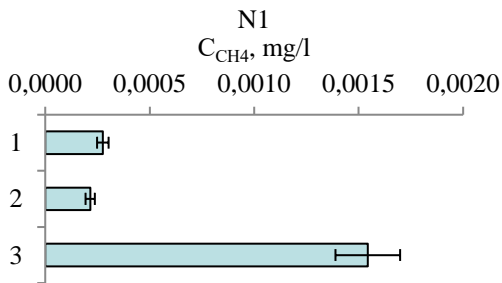
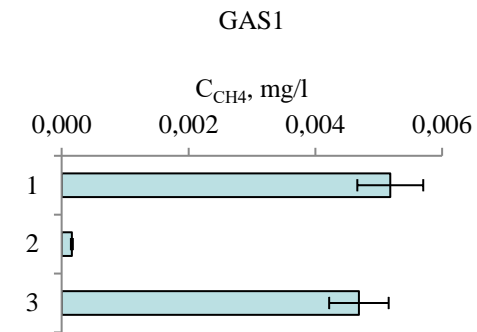
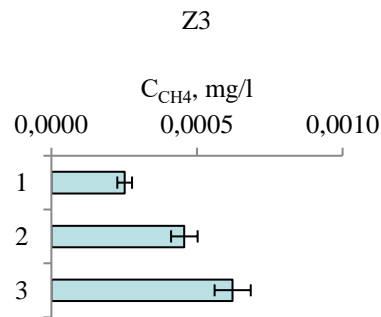
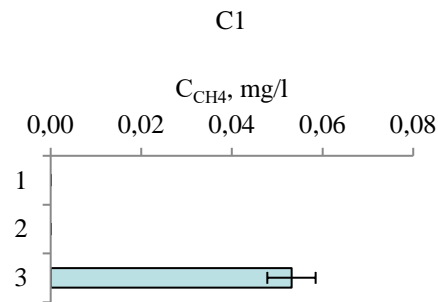
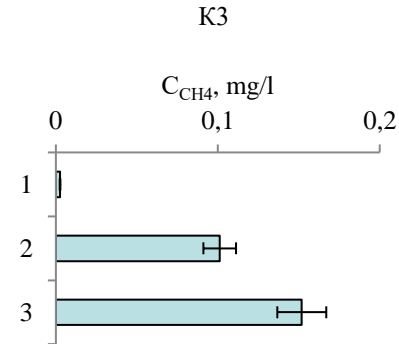
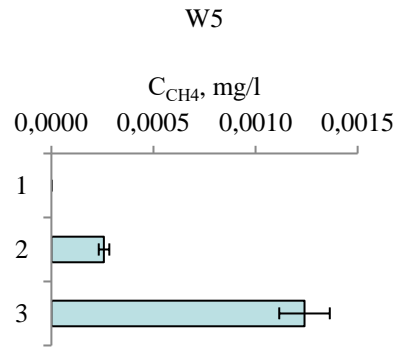
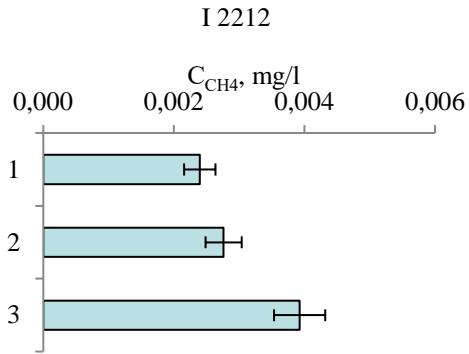
- CO_2 - Titrimetric method
- pH - Potentiometric method
- κ , T° - Conductometric method
- HCO_3^- - Titrimetric method
- Cl^- - Photometric method with mercury thiocyanate in the presence of iron (III) nitrate
- SO_4^{2-} - Photometric method with sulfonazo (III)
- Na, K - Flame photometric method
- Ca, Mg - Atomic absorption spectrometry method
- O_2 - Titrimetric determination according to Winkler
- Biochemical oxygen demand (BOD_5) - Titrimetric bottle method
- Chemical oxygen demand (COD) - Titrimetric Method
- Permanganate oxidizability (PO) - Titrimetric method
- Chromaticity - Photometric method
- NH_4^+ - Photometric method as Indophenol Blue
- NO_2^- - Photometric method with sulfanilamide and N-(1-naphthyl)ethylenediamine dihydrochloride
- NO_3^- - Photometric method with sulfanilamide and N-(1-naphthyl)ethylenediamine dihydrochloride after reduction in a cadmium reducer
- N_{total} - Photometric method with sample mineralization in a thermoreactor
- P_{min} , PO_4^{3-} - Photometric determination with ammonium molybdate, $\lambda = 882 \text{ nm}$

Oxygen regime of Onego Lake

Date	Lake District	Surface horizon				Near-bottom horizon			
		T, °C	O ₂ , mg/l	O ₂ , %	CO ₂ , mg/l	T, °C	O ₂ , mg/l	O ₂ , %	CO ₂ , mg/l
May 2020-2021	Average for the Lake	2.6-6.0	9.68-13.14 (11.56)	89-105	1.98-2.75 (2.22) *PB 3.19-4.384 (3.74)	2.5-5.2	10.57-13.13 (12.01)	80-96	1.98-2.97 (2.44) *PB 4.51-5.94 (4.45)
August 2020-2021	Deep sea part	15-20	8.45-9.56 (9.29)	92-102	1.76-3.10 (2.38) *K6-2.3	4.0-6.8	9.68-13.14 (10.78)	67-100	2.86-5.28 (4.43) *K6-8.91
	Shallow Bays	15-20	7.31-9.42 (8.78)	84-96	1.98-4.95 (3.40) *K3-4.6	10-17	7.28-11.54 (9.23)	72-94	2.75-8.25 (4.95) *K3-9.911

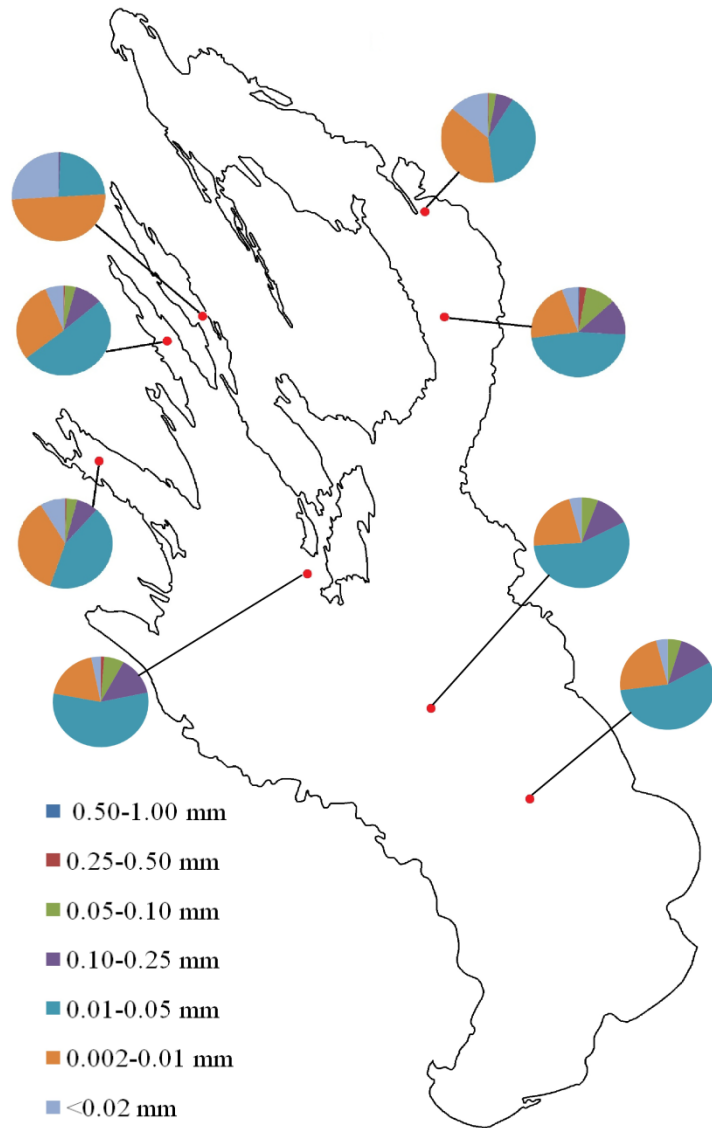
Results

Methane in water



Methane content in surface (1) and near-bottom water (2) - 100 cm, (3) - 30 cm at the bottom sediments of different stations of Onego Lake

Grain size of bottom sediments of Onego Lake



The grain size distribution of the study area is represented mainly by the fraction of dust particles (0.002-0.01 mm) from 53% to 86%.

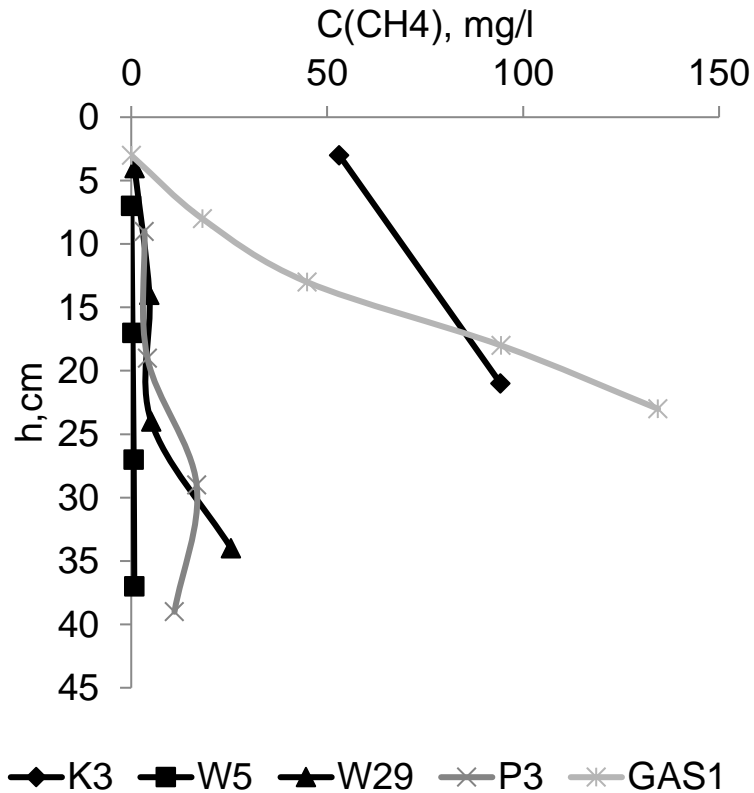
Content of clay particles (<0.002 mm) varies from 3% to 36%.

Fractions with the dimension of sand particles are not found in all bottom sediments and are mainly represented by fine particles (0.05-0.10 mm) from 0% to 24%.

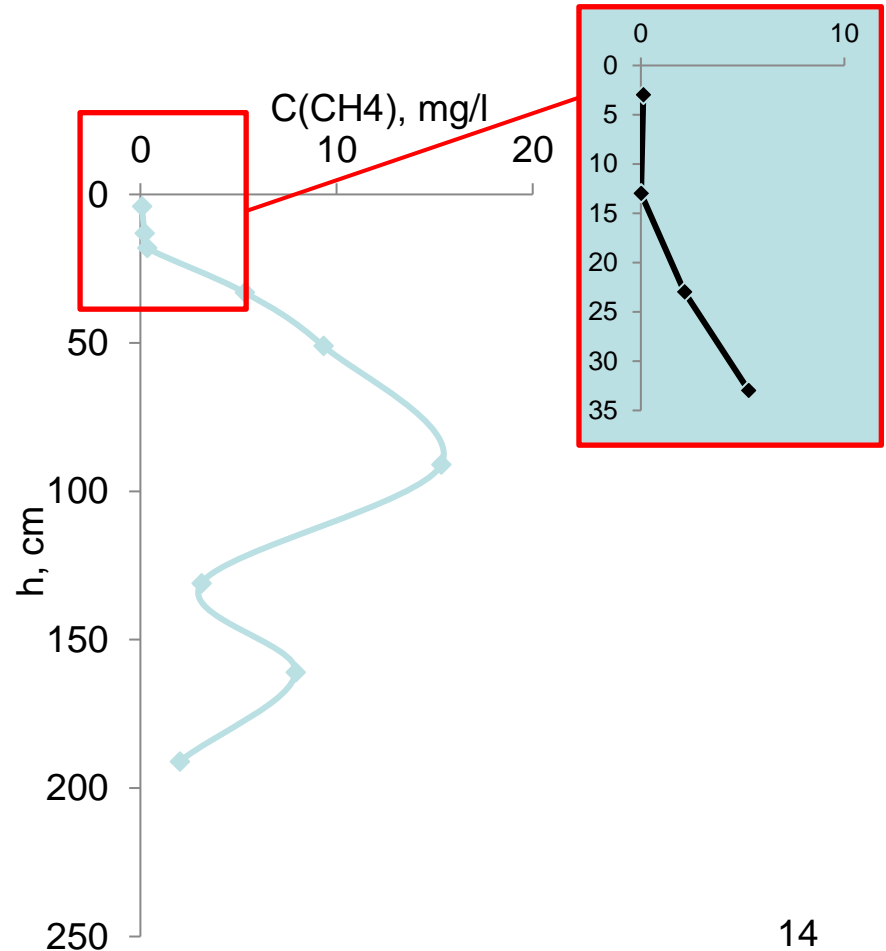
Results

Methane in Onego Lake bottom sediments

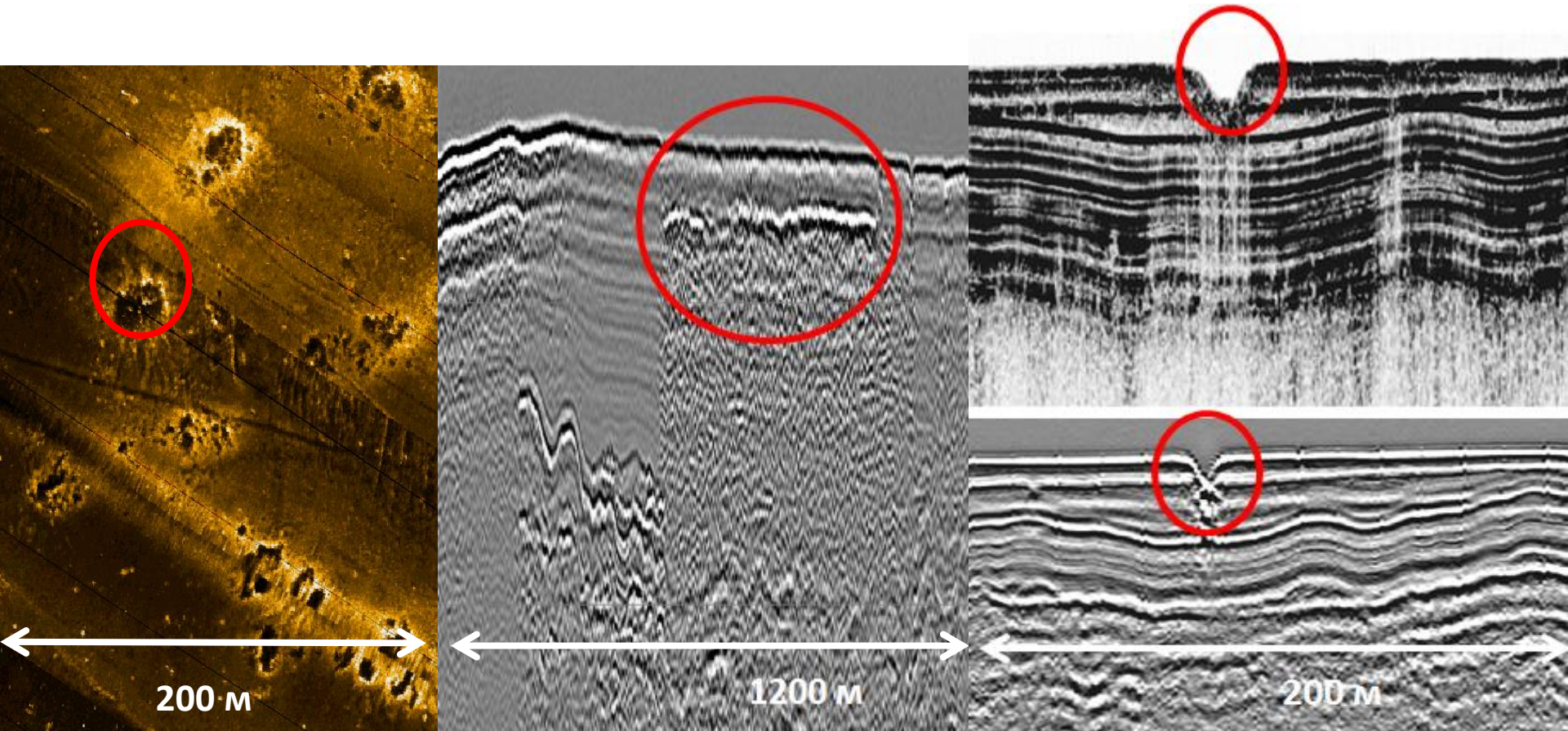
The concentration of methane in the bottom sediments of Onego Lake varied from 0.003 to 134.46 mg/l (average 13.14 mg/l)



Sampling with 3 m and 1 m samplers (station C1)



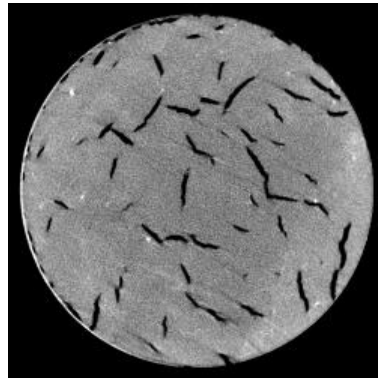
Geophysical research in the Petrozavodsk bay gas pockmarks



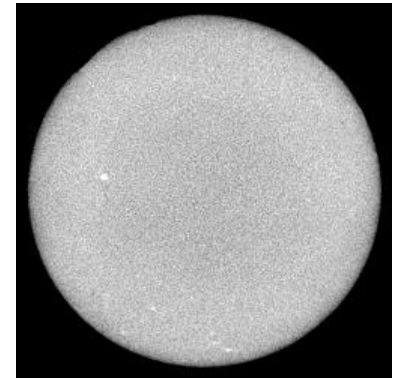
Subetto D., Rybalko A., Strakhovenko V., Belkina N., Tokarev M., Potakhin M., Aleshin M., Belyaev P., Dubois N., Kuznetsov V., Korost D., Loktev A., Shalaeva N., Kiskina A., Kostromina N., Kublitskiy Y., Orlov A. **Structure of Late Pleistocene and Holocene Sediments in the Petrozavodsk Bay, Lake Onego (NW Russia)** // Minerals, 2020. V. 10. P. 964
doi:10.3390/min10110964

Tomography of bottom sediments columns (vertical and horizontal sections)

Petrozavodsk Bay of Onego Lake (Rybalko et al. 2021)

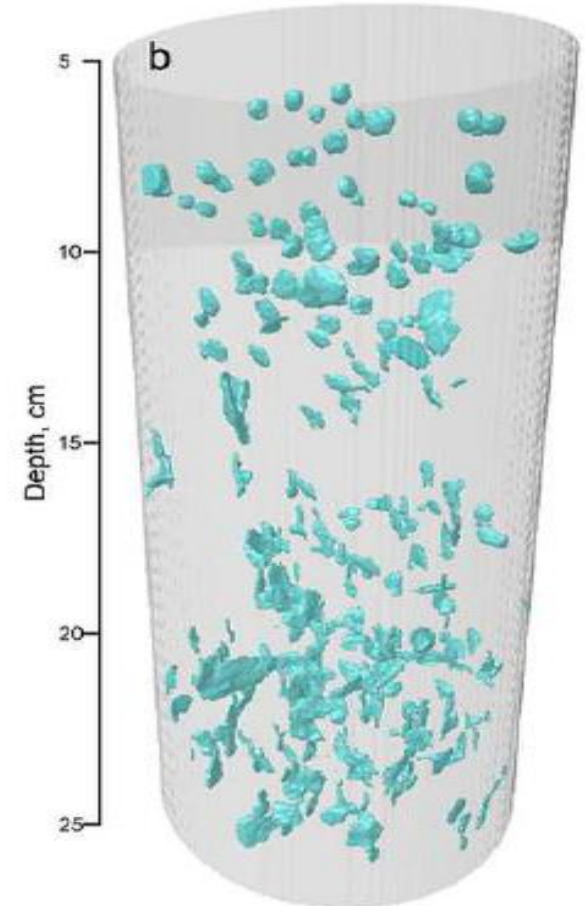
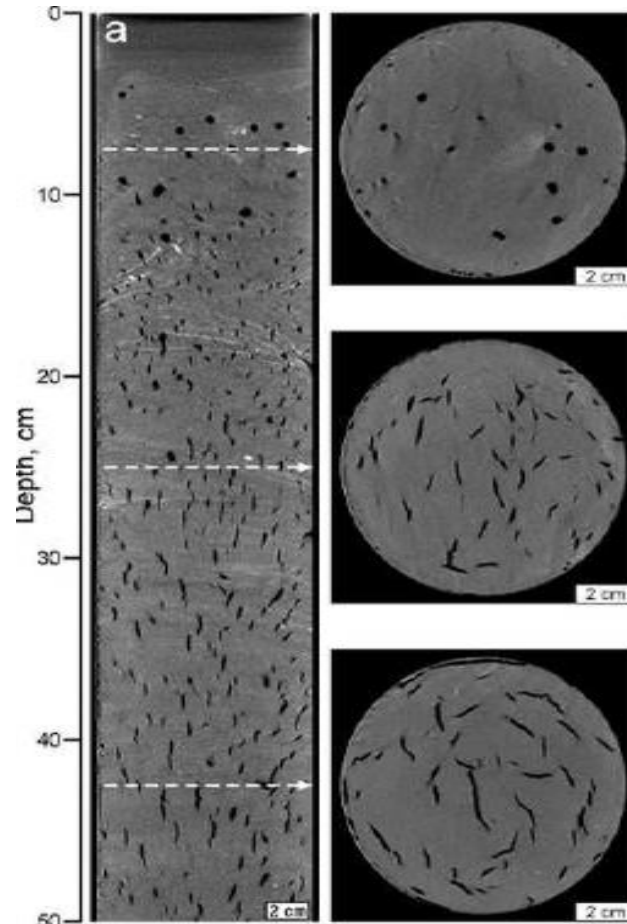
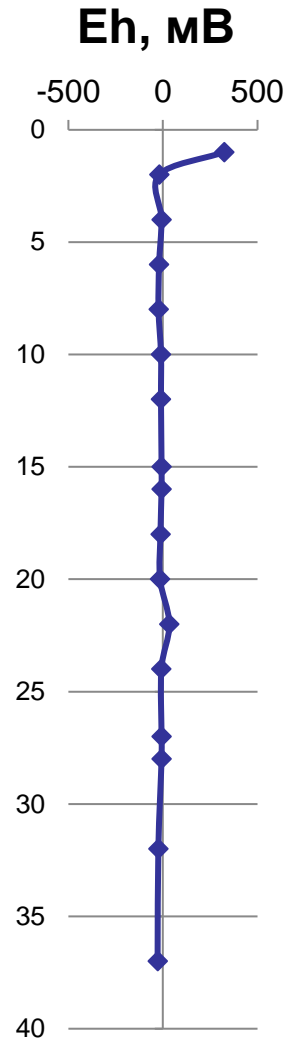


Porous structure of
bottom sediments –
as a result of gas
emanation



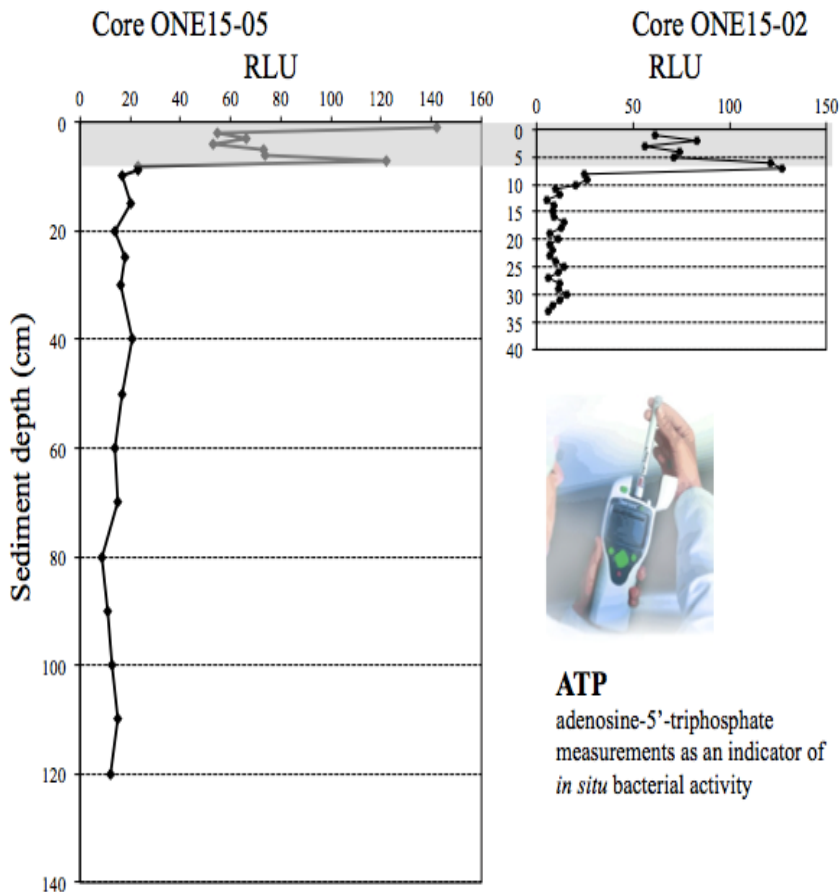
The usual structure of
bottom sediments

Anomaly of CH₄ in the sediments of the Petrozavodsk Bay



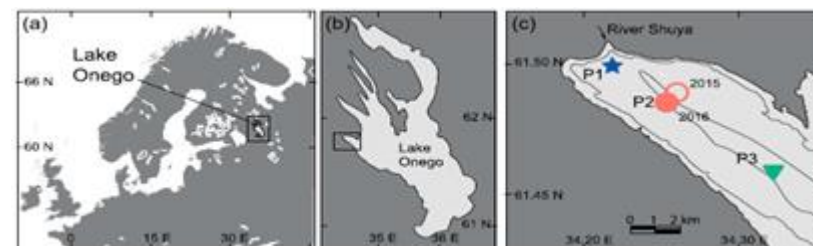
- a** - Vertical and separate horizontal radiographic contrast section views of the upper 50cm in core;
- b** – three-dimensional model of the cavity distribution (within the upper 25 cm)₁₇ resulting from degassing of the sediments.

Microbial activity in the sediments of Onego Lake



Thomas C., Frossard V., Perga M.-E., Tofield-Pasche N., Hofmann H., Dubois N., Belkina N., Zobkova M., Robert S., Lyautey E. Lateral variations and vertical structure of the microbial methane cycle in the sediment of Lake Onego (Russia) // *Inland Waters*, 2019. V. 9. I. 2: Life under Ice in Lake Onego (Russia) – An Interdisciplinary Winter Limnology Study. P. 205-226

[doi.:10.1080/20442041.2018.1500227](https://doi.org/10.1080/20442041.2018.1500227)

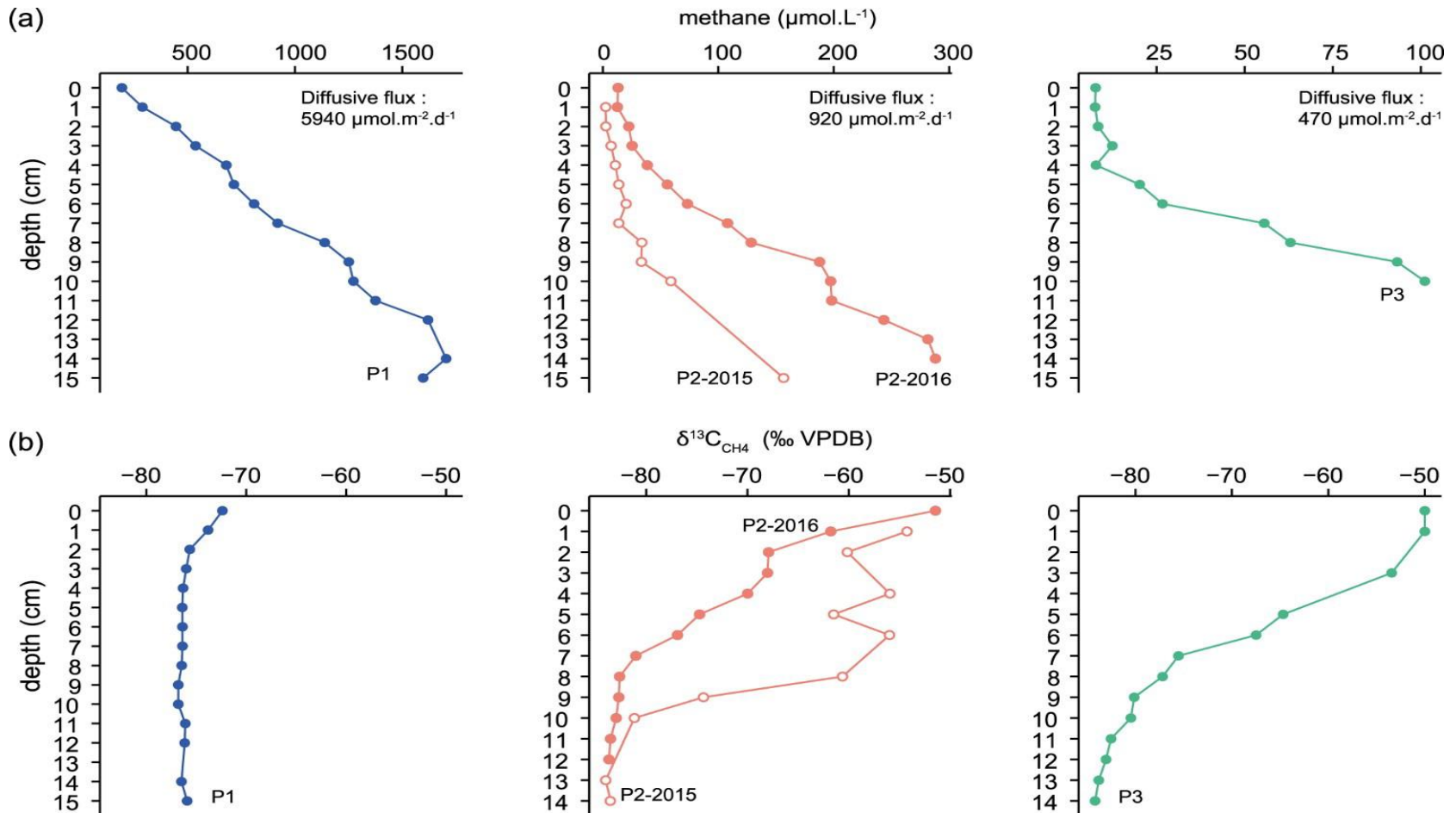


**Downcore record of microbial activity (ATP)
for Petrozavodsk bay sediments**

Microbial methane cycle in the sediments of Onego Lake

Family	1 cm	2 cm	3 cm	4 cm	5 cm	6 cm	7 cm	8 cm	9 cm	10 cm	15 cm
Aerobic methane oxydation											
Methylobacteriaceae	0.21	0.18	0.19	0.18	0.04	0.19	0.09	0.10	0.07	0.07	0.15
Methylophilaceae	2.94	2.67	1.89	2.69	1.46	2.15	1.78	1.26	0.23	0.13	0.22
Methylococcales CABC2E06	—	—	—	—	0.16	—	—	—	—	—	—
Methylococcales <i>Crenothrix</i>	0.15	0.15	0.21	0.25	0.04	0.25	0.26	0.09	0.01	0.04	0.06
Methylococcales <i>Methylobacter</i>	0.17	0.12	0.11	0.12	2.49	0.39	0.18	0.16	0.02	0.01	0.03
Methylococcales Methylococcaceae unknown genus	—	—	—	—	0.01	—	—	—	—	—	—
Methylococcales pIW-20	0.01	—	—	—	0.01	—	—	0.01	—	—	—
All Methylococcales	0 . 33	0 . 27	0 . 32	0 . 38	2 . 70	0 . 64	0 . 44	0 . 26	0 . 03	0 . 05	0 . 09
Anaerobic methane oxidation											
<i>Candidatus Methanoperedens</i>	—	0.02	0.03	0.05	0.03	0.71	0.61	0.20	0.04	0.06	0.48
<i>Candidatus Methylomirabilis</i>	1.11	0.89	1.49	1.35	0.63	3.33	1.83	1.03	0.46	0.56	0.29
total reads per sample	12 969	8435	13 607	28 428	51 894	14 810	12 526	17 947	17 704	20 779	22 577

Methane in the sediments of Lake Onego




Sedimentary profiles of (a) methane concentrations and (b) ratio of carbon isotopes from methane in Petrozavodsk Bay. Calculated methane diffusive flux.

Conclusions

1. The distribution of methane content in water and BS in different parts of Lake Onega is uneven. An increased concentration of methane in water was found in the Kondopoga Bay of Lake Onego. This is due to the active influx of organic matter into the water column and BS and its subsequent decomposition to substrates (H_2 , CO_2 , acetate, etc.), causing methanogenic archaea, both in BS and in anaerobic microniches formed in organic matter suspended in water. mineral particles. A high concentration of gas in water and BS were found in the Petrozavodsk Bay, in the places of detection of gas-saturated deposits – pockmarks.
2. The nature of the distribution of methane in the aquatic environment mainly determines its concentration in the upper horizons of BS, as evidenced by the presence of a close correlation in these two media, due to the pattern of diffusion flows at the BS-water boundary.

**The work was carried out within
the RFBR project No. 18-45-
100002.**



Thank you for your attention !